## Exercise 5.

• Conservation of Angular Momentum with RTT Eq. (3.16)

- Setting 
$$B_{\rm syst} = \widetilde{H}_0 = \int_{\rm syst} (\widetilde{r} \times \widetilde{v}) {\rm d}m$$
,  $\beta_{\rm syst} = \frac{{\rm d}\widetilde{H}_0}{{\rm d}m} = \widetilde{r} \times \widetilde{v}$ , gives:

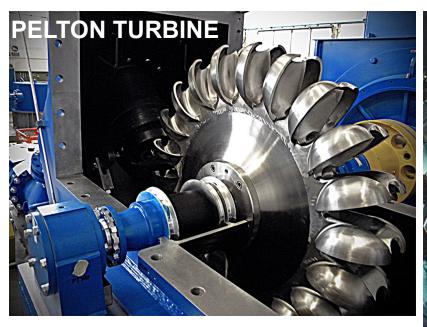
$$\frac{d\tilde{H}_0}{dt}\Big|_{\text{syst}} = \frac{d}{dt} \Big[ \int_{\text{CV}} (\tilde{r} \times \tilde{v}) \rho \, dV \Big] + \int_{\text{CS}} (\tilde{r} \times \tilde{v}) \rho (\tilde{v}_r \circ \tilde{n}) dA \qquad \text{Eq. (3.56)}$$

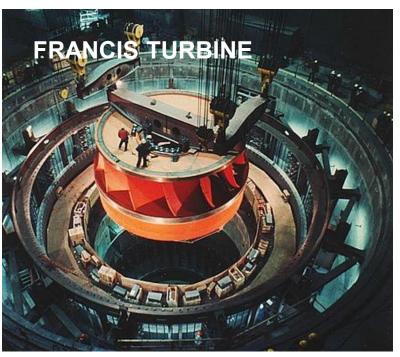
– Non-deformable and fix CV gives:

$$\sum \widetilde{M}_0 = \frac{\partial}{\partial t} \left[ \int_{\text{CV}} (\tilde{r} \times \tilde{v}) \rho \, dV \right] + \int_{\text{CS}} (\tilde{r} \times \tilde{v}) \rho (\tilde{v} \circ \tilde{n}) dA \qquad \text{Eq. (3.59)}$$

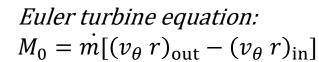
- Note the LHS remains if external momentum (force x distance) acts on the system:

$$\frac{\mathrm{d}\widetilde{H}_0}{\mathrm{d}t}\Big|_{\mathrm{Syst}} = \sum \widetilde{M}_0 = \sum (\widetilde{r} \times \widetilde{v})_0$$

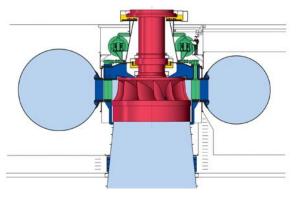


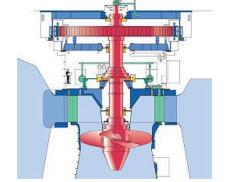






 $P = \omega M_0$ 





-Used in the initial design phase of all new turbines

## Exercise 5.

- Conservation of Energy with RTT Eq. (3.16)
  - Setting  $B_{\text{syst}} = E$ ,  $\beta_{\text{syst}} = \frac{dE}{dm} = e$ , which gives:

$$\frac{\mathrm{d}E}{\mathrm{d}t} = \frac{\mathrm{d}}{\mathrm{d}t} \left( \int_{\mathrm{CV}} e\rho \, \mathrm{d}V \right) + \int_{\mathrm{CS}} e\rho (\tilde{v}_r \circ \tilde{n}) \mathrm{d}A \qquad \text{Eq. (3.61)}$$

– Total energy (LHS) can be divided as:

$$\dot{E}_{\rm syst} = \dot{Q} - \dot{W}_{\rm shaft} - \dot{W}_{\rm visc} - \dot{W}_{\rm press}$$

- > The work done by pressure as:  $\dot{W}_{\rm press} = \int_{\rm CS} p \; (\tilde{v} \circ \tilde{n}) dA$  Eq. (3.63)
- The energy per unit mass as:

$$e = \hat{u} + 1/2 v^2 + gz$$
 Eq. (3.62)

- Adding the definition of enthalpy (sum of internal and pressure energy):  $h = \hat{u} + \frac{p}{a}$
- Non deformable and fix  $CV \rightarrow \frac{d}{dt} \Rightarrow \frac{\partial}{\partial t}$

## Exercise 5.

 Inserting all of this in the energy equation (3.61), move the pressure work to the RHS, and include the enthalpy h:

$$\dot{Q} - \dot{W}_{\rm S} - \dot{W}_{\rm V} = \frac{\partial}{\partial t} \left( \int_{\rm CV} \left( \widehat{u} + \frac{1}{2} v^2 + gz \right) \rho \, \mathrm{d}V \right) + \int_{\rm CS} \left( h + \frac{1}{2} v^2 + gz \right) \rho (\widetilde{v}_r \circ \widetilde{n}) \, \mathrm{d}A \qquad \text{Eq. (3.67)}$$

- Here:

*Q* = Heat transfer

 $\dot{W}_{\rm s}$ = Shaft work

 $\dot{W}_{\rm v}$ = Viscous (friction) work, often negligible

 $\hat{u}$  = Internal energy

v/2 = Kinetic energy

gz = Potential energy

h = Enthalpy

– Heat, Q, added or work, W, done by the system is positive!